

≡ Yellowcake

文A 27 languages ▾

Article [Talk](#)

Read Edit View history Tools ▾

From Wikipedia, the free encyclopedia



This article **needs additional citations for verification**. Please help [improve this article](#) by [adding citations to reliable sources](#). Unsourced material may be challenged and removed.

Find sources: "Yellowcake" – news · newspapers · books · scholar · JSTOR (March 2021) [\(Learn how and when to remove this message\)](#)

This article is about the chemical urania. For the type of cake, see [cake](#). For the falsified documents leading to an event popularly referred to as the 'yellowcake scandal', see [Niger uranium forgeries](#).

Not to be confused with [yellow substance](#).

Yellowcake (also called **uranium**) is a type of **uranium** concentrate **powder** obtained from **leach solutions**, in an intermediate step in the processing of **uranium ores**. It is a step in the processing of uranium after it has been mined but before fuel fabrication or **uranium enrichment**. Yellowcake concentrates are prepared by various extraction and refining methods, depending on the types of ores. Typically, yellowcakes are obtained through the milling and chemical processing of uranium **ore**, forming a coarse powder that has a pungent odor, is insoluble in water, and contains about 80% **uranium oxide**, which melts at approximately 2880 °C.

Overview [\[edit \]](#)

Originally raw uranium ore was extracted by traditional mining and this is still the case in many mines. It is first crushed to a fine powder by [passing it through crushers and grinders](#) to produce "pulped" ore. This is further processed with concentrated [acid](#), [alkaline](#), or [peroxide](#) solutions to leach out the uranium. However, nearly half of yellowcake production is now produced by [in situ leaching](#) in which the solution is pumped through the uranium deposit without disturbing the ground.^[1] Yellowcake is what remains after drying and filtering. The yellowcake produced by most modern mills is actually brown or black, not yellow; the name comes from the color and texture of the concentrates produced by early mining operations.^{[2][3]}

Yellowcake



Names	
Other names	urania
Identifiers	
CAS Number	1344-57-6
UNII	L70487KUZO
Properties	
Chemical formula	variable, see text
Appearance	Yellow granules (as Yellowcake); Brown or black granules (UO ₂ and others)
Melting point	2,880 °C (5,220 °F; 3,150 K)
Except where otherwise noted, data are given for materials in their standard state (at 25 °C [77 °F], 100 kPa).	
<div> verify (what is ?)</div>	
Infobox references	

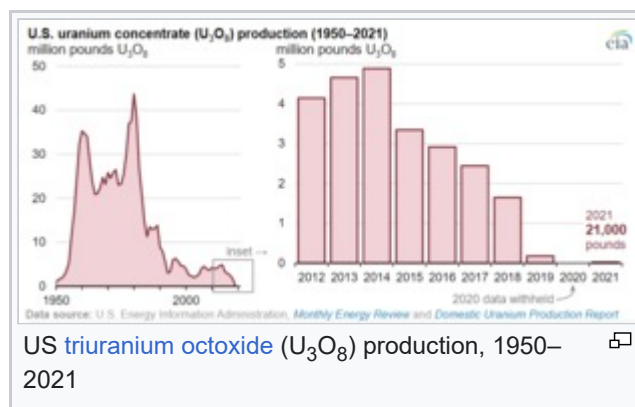


Initially, the compounds formed in yellowcakes were not identified; in 1970, the [U.S. Bureau of Mines](#) still referred to yellowcakes as the final precipitate formed in the milling process and considered it to be [ammonium diuranate](#) or [sodium diuranate](#). The compositions were variable and depended upon the leachant and subsequent precipitating conditions. The compounds identified in yellowcakes include [uranyl hydroxide](#), [uranyl sulfate](#), [sodium para-uranate](#), and [uranyl peroxide](#), along with various [uranium oxides](#). Modern yellowcake typically contains 70% to 90% [triuranium octoxide](#) (U_3O_8) by weight. Other oxides such as [uranium dioxide](#) (UO_2) and [uranium trioxide](#) (UO_3) exist.^[4]

Yellowcake is produced by all countries in which uranium ore is mined.^[1]



Yellowcake



Further processing [[edit](#)]

Yellowcake is used in the preparation of uranium fuel for [nuclear reactors](#), for which it is smelted into purified UO_2 for use in [fuel rods](#) for [pressurized heavy-water reactors](#) and other systems that use natural [unenriched uranium](#).

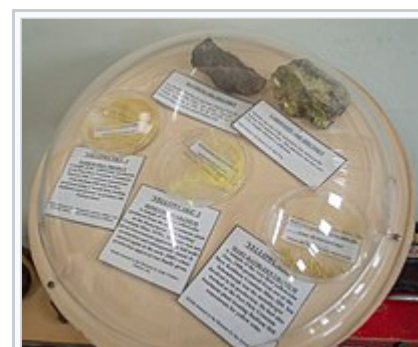
Purified uranium can also be [enriched into the isotope U-235](#). In this process, the uranium oxides are combined with [fluorine](#) to form [uranium hexafluoride](#) gas (UF_6). Next, the gas undergoes [isotope separation](#) through the process of [gaseous diffusion](#), or in a [gas centrifuge](#). This can produce [low-enriched uranium](#) containing up to 20% U-235 that is suitable for use in most large civilian electric-power reactors. With further processing, one obtains [highly enriched uranium](#), containing 20% or more U-235, that is suitable for use in compact nuclear reactors—usually used to power naval warships and [submarines](#). Further processing can yield [weapons-grade](#) uranium with U-235 levels usually above 90%, suitable for [nuclear weapons](#).

Radioactivity and safety [[edit](#)]

The uranium in yellowcake is almost exclusively (>99%) [U-238](#), with very low radioactivity. U-238 has a [half-life](#) of 4.468 billion years and emits radiation at a slow rate. This stage of processing is before the more radioactive U-235 is concentrated, so by definition, this stage of uranium has the same radioactivity as it did in nature when it was underground, as the proportions of isotopes are at their native relative concentration. Yellowcake is hazardous when inhaled.^[5]

See also [[edit](#)]

- [Uranium ore deposits](#)
- [Uranium mining](#)
- [Uraninite](#), an ore that is mostly [uranium dioxide](#) (UO_2)
- [Yellowcake forgery](#), fraudulently depicted Saddam Hussein trying to buy uranium powder
- [Sequoyah Fuels Corporation](#), an American company involved in yellowcake processing



Yellowcake and ore

- **COMINAK**, a **Niger** uranium mining and processing company
- **SOMAIR**, a Niger uranium mining and processing company
- **Vanadium(V) oxide**, hydrous precipitates of which are known as "redcake"

References [\[edit \]](#)

1. ^a ^b Gil, Laura (2018), "[Uranium leaching: How yellowcake is made](#)" , IAEA Bulletin (Online), vol. 59, iss. 2, pp. 22-23.
2. ^a "[Yellowcake](#)" . U.S. Nuclear Regulatory Commission. Retrieved 12 April 2014.
3. ^a "[Yellowcake](#)" . *European Nuclear Society nuclear glossary*. Archived from [the original](#)  on 6 July 2017. Retrieved 10 July 2017.
4. ^a Hausen, Donald M. (1998). "Characterizing and Classifying Uranium Yellow Cakes: A Background". *JOM*. **50** (12): 45–47. [Bibcode:1998JOM....50I..45H](#) . doi:[10.1007/s11837-998-0307-5](#) . [S2CID 97023067](#) .
5. ^a Keith, Sam; Faroon, Obaid; Roney, Nickolette; Scinicariello, Franco; Wilbur, Sharon; Ingerman, Lisa; Llados, Fernando; Plewak, Daneil; Wohlers, David; Diamond, Gary (February 2013). [Health Effects](#) . Agency for Toxic Substances and Disease Registry (US). Retrieved 22 August 2021.

V · T · E

Uranium compounds

[show]

Categories: Uranium compounds | Oxides | Nuclear materials

This page was last edited on 24 May 2024, at 19:34 (UTC).

Text is available under the [Creative Commons Attribution-ShareAlike License 4.0](#); additional terms may apply. By using this site, you agree to the [Terms of Use](#) and [Privacy Policy](#). Wikipedia® is a registered trademark of the [Wikimedia Foundation, Inc.](#), a non-profit organization.

[Privacy policy](#) [About Wikipedia](#) [Disclaimers](#) [Contact Wikipedia](#) [Code of Conduct](#) [Developers](#) [Statistics](#)

[Cookie statement](#) [Mobile view](#)

